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UTILITY PATENT APPLICATION TRANSMITTAL

Attorney Docket No. 125/116

First Inventor or Application Identified Brady et al.

Title IMPROVEMENTS IN AND RELATING TO CORROSIGN

Only for new nonprovisional applications under 37 C.F.R. § 1.53(b)) Express Mail Label No. EJ227695956US Assistant Commissioner for Patents **APPLICATION ELEMENTS** ADDRESS TO: **Box Patent Application** See MPEP chapter 600 concerning utility patent application contents. Washington, DC 20231 * Fee Transmittal Form (e.g., PTO/SB/17) Microfiche Computer Program (Appendix) (Submit an original and a duplicate for fee processing) Nucleotide and/or Amino Acid Sequence Submission X Specification Total Pages (if applicable, all necessary) (preferred arrangement set forth below) Computer Readable Copy - Descriptive title of the Invention - Cross References to Related Applications Paper Copy (identical to computer copy) - Statement Regarding Fed sponsored R & D Statement verifying identity of above copies - Reference to Microfiche Appendix ACCOMPANYING APPLICATION PARTS - Background of the Invention - Brief Summary of the Invention X Assignment Papers (cover sheet & document(s)) - Brief Description of the Drawings (if filed) 37 C.F.R.§3.73(b) Statement [Power of - Detailed Description Attorney (when there is an assignee) - Claim(s) 9 English Translation Document (if applicable) - Abstract of the Disclosure Copies of IDS Information Disclosure 10 Drawing(s) (35 U.S.C. 113) Statement (IDS)/PTO-1449 Total Sheets Citations 3. Preliminary Amendment 4. Oath or Declaration [Total Pages 3 Return Receipt Postcard (MPEP 503) X Newly executed (original or copy) 12. (Should be specifically itemized) Copy from a prior application (37 C.F.R. § 1.63(d)) * Small Entity Statement filed in prior application (for continuation/divisional with Box 16 completed) 13. Statement(s) Status still proper and desired (PTO/SB/09-12) **DELETION OF INVENTOR(S)** Certified Copy of Priority Document(s) Signed statement attached deleting 14. (if foreign priority is claimed) inventor(s) named in the prior application, see 37 C.F.R. §§ 1.63(d)(2) and 1.33(b). 5. Other: NOTE FOR ITEMS 1 & 13 IN ORDER TO BE ENTITLED TO PAY SMALL ENTITY FEES, A SMALL ENTITY STATEMENT IS REQUIRED (37 C.F.R. § 1.27), EXCEPT IF ONE FILED IN A PRIOR APPLICATION IS RELIED UPON (37 C.F.R. § 1.28). 16. If a CONTINUING APPLICATION, check appropriate box, and supply the requisite information below and in a preliminary amendment: of prior application No: Continuation-in-part (CIP) Divisional Continuation Group / Art Unit: Examiner Prior application information: For CONTINUATION or DIVISIONAL APPS only: The entire disclosure of the prior application, from which an oath or declaration is supplied under Box 4b, is considered a part of the disclosure of the accompanying continuation or divisional application and is hereby incorporated by reference. The incorporation can only be relied upon when a portion has been inadvertently omitted from the submitted application parts. 17. CORRESPONDENCE ADDRESS X Correspondence address below Customer Number or Bar Code Labe ! (Insert Customer No. or Attach bar code label here) Richard E. Jenkins Name JENKINS & WILSON, P.A. Suite 1400 University Tower Address 3100 Tower Boulevard Zip Code 27707 NC State City Durham 001-919-419-0383 001-919-493-8000 Fax Telephone USA Country Registration No (Attorney/Agent) 28,428 Name (Print/Type) Richard E. Jenkins Sianature

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property (times number of properties)

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SUBTOTAL (3)

For each additional invention to be examined (37 CFR § 1.129(b))

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Complete if Known

TRANSMITTAL

Fee Description

Independent claims in excess of 3

Multiple dependent claim, if not paid

** Reissue independent claims over original patent

SUBTOTAL (2) (\$) 120.00

** Reissue claims in excess of 20 and over original patent

Claims in excess of 20

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	Richard E. Jenkins	Registration No. (Attorney/Agent)	28,428	Telephone	001-919-493-8000
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September 22, 2000

"Express Mail" mailing number T227695665

Date of Deposit 22 September 2000

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Lillian S. Glenn

Commissioner for Patents BOX PATENT APPLICATION Washington, D.C. 20231

Re:

U.S. Patent Application for IMPROVEMENTS IN AND RELATING TO CORROSION INHIBITING FORMULATIONS which claims priority to UK 9924358.6 filed 10/14/99

Our File No. 125/116

Sir:

Please find enclosed the following:

- A U.S. patent application for IMPROVEMENTS IN AND RELATING TO CORROSION INHIBITING FORMULATIONS (35 pages);
- Preliminary Amendment (2 pages);
- An executed Declaration (3 pages);
- 4. An unexecuted Small Entity Statement (2 pages);
- 5. An executed Assignment (2 pages);
- 6. Recordation Form Cover Sheet (PTO-1595; 1 page);
- Utility Patent Application Transmittal (Form PTO/SB/05; 1 page);
- 8. Fee Transmittal (Form PTO/SB/17; 1 page) in duplicate;

Commissioner for Patents September 22, 2000 Page 2

- 9. A check in the amount of \$505.00 to cover the small entity application filing fee (\$465.00) and the assignment recordation fee (\$40.00);
- 10. A return-receipt postcard to be returned to our offices with the U.S. Patent and Trademark date stamp thereon; and
- 11. A Certificate of Express Mail No.: EJ227695956US.

Please contact our offices if there are any questions with respect to this matter.

Respectfully submitted,

JENKINS & WILSON, P.A.

Richard E. Jenkins

Registration No.: 41,085

REJ/lsg

Enclosures

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of:

Brady, Arthur Albert; Clubley, Brian George

Application No.:

Filed on:

Title:

IMPROVEMENTS IN AND RELATING TO CORROSION INHIBITING

FORMULATIONS

STATEMENT CLAIMING SMALL ENTITY STATUS (37 CFR 1.9(f) and 1.27(b)--SMALL BUSINESS CONCERN

I hereby state that I am an official of the small business concern empowered to act on behalf of the concern identified below:

BRAD-CHEM TECHNOLOGY LIMITED Unit C4, Moss Industrial Estate Leigh, Lancs WN7 3PT UNITED KINGDOM

I hereby state that the above identified small business concern qualifies as a small business concern, as defined in 13 CFR 121.12, and reproduced in 37 CFR 1.9(d), for purposes of paying reduced fees to the United States Patent and Trademark Office under Sections 41(a) and (b) of Title 35, United States Code, in that the number of employees of the concern, including those of its affiliates, does not exceed 500 persons. For purposes of this statement, (1) the number of employees of the business concern is the average over the previous fiscal year of the concern of the persons employed on a full-time, part-time or temporary basis during each of the pay periods of the fiscal year, and (2) concerns are affiliates of each other when either, directly or indirectly, one concern controls or has the power to control the other, or a third-party or parties controls or has the power to control both.

I hereby state that rights under contract or law have been conveyed to, and remain with, the small business concern identified above, with regard to the invention described in the specification filed herewith, with title as listed above.

If the rights held by the above identified small business concern are not exclusive, each individual, concern or organization having rights in the invention is listed below* and no rights to the invention are held by any person, other than the inventor, who would not qualify as an independent inventor under 37 CFR 1.9(c), if that person made the invention, or by any concern which would not qualify as a small business concern under 37 CFR 1.9(d) or a nonprofit organization under 37 CFR 1.9(e).

No such person, concern, or organization exists.

I acknowledge the duty to file, in this application or patent, notification of any change in status resulting in loss of entitlement to small entity status prior to paying, or at the time of paying, the earliest of

(Small Entity-Small Business--page 1 of 2)

the time that that that and relations to the total and the time that that

the issue fee or any maintenance fee due after the date on which status as a small business entity is no longer appropriate. (37 CFR 1.28(b))

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further, that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application, any patent issuing thereon, or any patent to which this verified statement is directed..

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Lillian S. Glenn

PATENT

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Application of: Brady et al.

Group Art Unit:

Serial No.:

Examiner:

Filed:

Docket No.: 125/116

For: IMPROVEMENTS IN AND RELATING TO CORROSION INHIBITING FORMULATIONS

PRELIMINARY AMENDMENT

Commissioner for Patents BOX PATENT APPLICATION Washington, D.C. 20231

Sir:

AMENDMENTS

Kindly amend the subject application as follows:

IN THE CLAIMS:

Claim 10, lines 2 and 3, please delete the phrase "any one of the preceding claims" and insert in place thereof --claim 1--.

Claim 27, lines 1 and 2, please delete the phrase "any one of the preceding claims" and insert in place thereof --claim 1--.

Claim 29, lines 2 and 3, please delete the phrase "any one of claims 1-9" and insert in place thereof --claim 1--.

REMARKS

The amendments to the claims as set forth above are intended to remove all multiple dependent claims from the subject application and to more particularly point out and distinctly claim the subject invention.

Respectfully submitted,

JENKINS & WILSON, P.A.

Richard E. Jenkins

Reg. No.: 28,428

Suite 1400 University Tower 3100 Tower Boulevard Durham, North Carolina 27707 Telephone: (919) 493-8000

125/116 REJ/lsg

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IMPROVEMENTS IN AND RELATING TO CORROSION INHIBITING FORMULATIONS

Field of the Invention

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The present invention relates to improvements in and relating to corrosion inhibiting formulations.

Background of the Invention

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A wide variety of additives have been developed and are in commercial use for the protection of metals against corrosion in organic or aqueous media.

Amongst ferrous metal corrosion inhibitors, carboxylic acids are widely used, for example, those with long chain alkyl or aryl groups which impart solubility to the carboxylic acids in organic systems, for example hydrocarbon oils. Particularly effective compounds are acyl amino acids of structure (I):

$$R^{1}-C(0)-N(R^{2})-CH_{2}-CO_{2}H$$

(I)

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where R^1 is a long chain alkyl group and R^2 is hydrogen or methyl.

Where water solubility is required, it is common practice to employ a water soluble salt of the carboxylic acid, for example, a sodium salt or a trialkanolamine salt.

Amongst non-ferrous metal corrosion inhibitors (also known as metal passivators or deactivators), the most widely used compounds for protection of copper and copper alloys are triazole derivatives, particularly benzotriazole or tolutriazole. As above, derivatives are well known which enhance solubility in organic hydrocarbons or water. For example structures of formula (II) are in commercial use:

 $T^1-CH_2-NR^3_2$

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(II)

where T^1 is a triazole derivative, R^3 is hydroxyalkyl (for water solubility) or alkyl (for oil solubility).

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When ferrous metal and non-ferrous metal corrosion inhibition are both required in an oil-based medium two oil-soluble corrosion inhibitors are used. Similarly, when both ferrous metal and non-ferrous metal corrosion inhibitors are required in an aqueous medium, two water-soluble corrosion inhibitors are used.

However, in a number of applications, solubility of the corrosion inhibitor in both organic hydrocarbons and water is desirable. Such applications include, but are not limited to, those involving emulsions or those where water may be present as a contaminant of oil systems and vice versa. Applications where dual solubility would be of utility include, but are not limited to, industrial oils, crankcase lubricants, metalworking fluids, hydraulic fluids, rolling oils, brake fluids, fuel systems, surface coatings and water-soluble paints.

In an effort to reduce the amount of organic hydrocarbon material used in many technologies, aqueous emulsion formulations are being developed so that the range of technical areas in which emulsions will have importance is increasing. However, the conventional corrosion inhibition compounds described above result in either organic solubility or water solubility.

It is an aim of preferred embodiments of the present invention to provide an improved corrosion inhibitor.

Summary of the Invention

According to a first aspect of the present invention there
is provided a formulation comprising:

a) an acyl amino acid derivative of the formula (III):

$$R^4 - (CO) - N(R^5) - CH_2 - CO_2H$$

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(III)

in which

 \mathbb{R}^4 is C_8 - C_{30} optionally substituted alkyl

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and R^5 is hydrogen or methyl, and

b) a N,N'-disubstituted aminomethyl triazole derivative of the formula (IV):

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$$T^2$$
-CH₂-NR⁶₂

in which

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 ${\tt T^2}$ is an optionally substituted 1,2,3-benzotriazole group, or an optionally substituted 1,2,4-triazole group, and

R⁶ is a hydroxyalkyl group.

It has surprisingly been found that formulations in accordance with the present invention are soluble in both oil and water and give corrosion protection to a range of metals.

Suitably, the acyl amino acid derivative of formula (III) is organic hydrocarbon-soluble.

Suitably, the N,N'-disubstituted aminomethyl triazole derivative of formula (IV) is water-soluble.

Generally, unless otherwise stated herein or unless the
context requires otherwise, any alkyl may be either a
straight or branched chain alkyl. Except where otherwise
stated in this specification optional substituents of an
alkyl group may include, aryl, alkenyl, or alkoxy ether
groups. Except where otherwise stated in this
specification the alkyl group may optionally be
interrupted, for example, by an unsaturated linkage and/or
an oxygen. Each alkyl group may comprise one or more
substituents. Suitably, alkyl groups are unsubstituted.

Preferably, R^4 is $C_{10}-C_{20}$ alkyl. More preferably R^4 is $C_{11}C_{17}$ alkyl. Especially preferred R^4 groups are $C_{11}H_{23}$ derived from lauric acid and $C_{17}H_{33}$ derived from oleic acid.

Preferably R⁵ is methyl. Preferably, R⁵ is an unsubstituted methyl group.

- Generally, unless otherwise stated herein or unless the context requires otherwise, any benzotriazole group may be optionally substituted by one or more alkyl groups.

 Suitable substituents include C₁-C₄ alkyl groups. A benzotriazole group may comprise four substituents or
- less. Suitably, a benzotriazole group comprises three substituents or less, preferably two substituents or less and more preferably one substituent or less. Suitably, a benzotriazole group is unsubstituted. Suitably, alkyl group substituents of benzotriazole groups are unsubstituted.

Preferably, T^2 is a 1,2,3-benzotriazole group. More preferably T^2 is a benzotriazole or tolutriazole.

- Suitably, R⁶ is a hydroxyalkyl group containing from 1-4 carbon atoms. Suitably, R⁶ is a hydroxyalkyl group containing from 1-4 hydroxyl groups. Suitably, the hydroxyalkyl group comprises an unsubstituted hydroxyalkyl group. Preferably, R⁶ is 2-hydroxyethyl or hydroxypropyl most preferably R⁶ is 2-hydroxyethyl. The R⁶ groups of formula (IV) may comprise the same or different hydroxyalkyl group.
- Suitably, the formulation is soluble in organic

 hydrocarbons. Suitably the formulation is soluble in
 water. Preferably, the formulation is soluble in both
 organic hydrocarbons and water.

Generally, unless otherwise stated herein or otherwise required by the context, a formulation is soluble in organic hydrocarbons if at 25°C and 1 atmosphere pressure it has a solubility of at least 0.000001wt%, suitably at least 0.0001wt%, preferably at least 0.01wt%, more preferably at least 0.5wt% and particularly at least 1.0wt% in Naphthenic oil. The naphthenic oil is suitably Nynas T22 (trade mark) available from Nynas Naphthenics Ltd.

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Generally unless otherwise stated herein or otherwise required by the context, a formulation is soluble in organic hydrocarbons if at 25°C and 1 atmosphere pressure it has a solubility of at least 0.000001wt%, suitably at least 0.0001wt%, preferably at least 0.01wt%, more preferably at least 0.5wt% and particularly at least 1.0wt% in toluene.

Generally, unless otherwise stated herein or otherwise

required by the context, a formulation is soluble in water
if at 25°C and 1 atmosphere pressure it has a solubility
of at least 0.000001wt%, suitably at least 0.0001wt%,
preferably at least 0.01wt%, more preferably at least
0.5wt% and particularly at least 1.0wt% in water. The

water used to determine solubility is suitably laboratory
grade de-ionised water.

Solubility should be tested by preparing a mixture of the appropriate concentration of formulation to solvent,

warming the mixture, with stirring, at 40-50°C for approximately 15 minutes. The mixture should then be left to stand overnight at ambient temperature and then

visually inspected to ascertain whether or not complete solution has been obtained.

The formulation may comprise more than one acyl amino acid derivative of the formula (III). The formulation may comprise more than one N,N'-disubstituted aminomethyl triazole derivative of the formula (IV).

Especially preferred mixtures are those from combination
of N-oleoyl sarcosine and bis (2-hydroxyethyl) amino
methyl tolutriazole, N-oleoyl sarcosine and bis (2hydroxyethyl) amino methyl benzotriazole, N-lauroyl
sarcosine and bis (2-hydroxyethyl) amino methyl
tolultriazole and N-lauroyl sarcosine and (2-hydroxyethyl)
amino methyl benzotriazole.

The optimum ratio of compound (III) to compound (IV) will depend on the degree of solubility required in each solvent. For example, higher levels of compound (III)

will generally increase solubility in organic hydrocarbons and higher levels of compound (IV) will generally increase solubility in water.

The formulation may comprise less than 99wt%, less than 95%, less that 80wt% or less than 55wt% of the acyl amino acid of formula (III).

The formulation may comprise greater than 1wt%, greater than 5wt%, greater than 20wt% and greater than 45wt% of the acyl amino acid of formula (III).

The formulation may comprise less than 99wt%, less than 95wt%, less than 80%wt, less than 55wt% of the N,N'-disubstituted aminomethyl triazole of formula (IV).

- The formulation may comprise greater than lwt%, greater than 5wt%, greater than 20wt% and greater than 45wt% of the N,N'-disubstituted aminomethyl triazole of formula (IV).
- The mole ratio of compound (III) to compound (IV) is suitably between 1:0.2 and 1:2, preferably between 1:0.5 and 1:1 and more preferably between 1:0.6 and 1:0.9.
- The formulation may further comprise minor amounts of additional additives. Suitably, the formulation comprises less than 10wt%, preferably less than 5wt% and more preferably less than 1wt% in total of additional additives.

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Non-limiting examples of additional additives are:

phenolic or aminic antioxidants, further corrosion or rust
inhibitors, further metal deactivators, extreme

25 pressure/antiwear additives, viscosity index improvers,
 pour point depressants, dispersants/surfactants,
 antifoams, biocides, complexing agents.

Non-limiting examples of suitable phenolic antioxidants 30 are:

2,6-di-t-butyl phenol derivatives, 2-t-butyl-6-methyl phenol derivatives.

Non-limiting examples of suitable aminic antioxidants are:

alkylated diphenylamine derivatives and alkylated anaphthylamine derivatives.

Non-limiting examples of suitable corrosion inhibitors/rust inhibitors are:

alkyl or aryl mono, di or poly carboxylic acids and their ammonium salts, amine alkanolamine or metal salts, alkenyl succinic acid and derivatives, aryl sulphonic acid salts, phosphoric acid, poly phosphoric acids and derivatives, phosphoric acids, fatty acid alkanolamides, imidazoline derivatives.

Non-limiting examples of suitable metal de-activators are:

1,2,4-triazole derivatives, benzotriazole and alkylated
20 benzotriazole derivatives, mercaptobenzothiazole and its
sodium salt.

Non-limiting examples of suitable extreme pressure/antiwear additives are:

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zinc dialkyl dithiophosphates, amine salts of substituted phosphoric acid derivatives, triphenyl phosphorothicate and alkylated derivatives, molydenum dithiocarbamate derivatives, tri-aryl phosphates, sulphurised hydrocarbons e.g. vegetable oils.

Non limiting examples of suitably viscosity index improvers/pour depressants are:

polyacrylates, polymethacrylates, clefin co-polymers, styrene-acrylate co-polymers, alkylated naphthalene derivatives.

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Non-limiting examples of suitable dispersants/surfactants are:

polybutenyl succinic acid amides, basic magnesium, calcium 10 and barium sulphonates and phenolates.

Non-limiting examples of suitable antifoams are:

polysiloxanes, ethylene/propylene oxide co-polymers.

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Non limiting examples of suitable biocides are:

isothiazolone derivatives, boron amide derivatives.

- 20 Non-limiting examples of suitable complexing agents are:
 - ethylene diamine tetra-acetic acid and derivatives, citric acid.
- Compounds (III) and compounds (IV) are commercially available or may be prepared by conventional methods as described in the scientific literature. For example, compounds (III) may be prepared by reaction of an appropriate acyl chloride with an appropriate amino acid.
- 30 Compounds (IV) may be prepared by reacting an appropriate triazole derivative with formaldehyde and a hydroxylcontaining secondary amine by the Mannich reaction.

According to a second aspect of the invention there is provided a composition comprising:

- a) a formulation according to the first aspect of the invention and
 - b) a diluent.
- 10 Suitable diluents include water, organic hydrocarbons or a mixture thereof.

Suitable organic hydrocarbons include natural or synthetic aliphatic or aromatic compounds of carbon and hydrogen, optionally containing unsaturated linkages, ester groups or hetero atoms e.g. oxygen. Non-limiting examples of suitably organic hydrocarbons are:

octane, kerosine, white spirit, petroleum-based
hydrocarbons such as naphthenic oils or paraffinic oils,
vegetable oils, synthetic carboxylic acid ester, phosphate
esters, poly α olefins, poly isobutylenes, alkylated
aromatic hydrocarbons, ethylene glycol, propylene glycol,
polyalkylene glycols, glycol ethers.

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The water according to the second aspect of the invention may be distilled water, de-ionised water, natural water or synthetic hard water, for example.

30 Preferably, the diluent comprises one or more of water, glycols or mineral oils. More preferably, the diluent comprises propylene glycol and/or naphthenic oil.

The composition of the second aspect of the invention advantageously provides the formulation of the first aspect of the invention in a form which may facilitate use and/or handling thereof.

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The composition of the second aspect of the invention may be manufactured by mixing the formulation of the first aspect of the invention with the diluent. Alternatively, or in addition, the composition of the second aspect of the invention may be provided by preparing the formulation of the first aspect in the presence of the diluent.

The composition according to the second aspect of the invention may be further diluted in use, by addition of any of the diluents listed above.

The composition according to the second aspect of the invention may comprise at least 5wt% and suitably at least 20wt% of the formulation according to the first aspect.

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The composition according to the second aspect of the invention may comprise less than 95wt%, suitably less than 90wt% of the formulation according to the first aspect.

The composition according to the second aspect of the invention may comprise at least 5wt% and suitably at least 20wt% diluent.

The composition according to the second aspect of the invention may comprise less than 95wt% and suitably less than 90wt% diluent.

The composition may comprise further additives. Suitably, the composition may comprise 5wt%, or 3wt% or 1wt% or less of further additives.

According to a third aspect of the present invention there is provided a composition in accordance with the second aspect of the invention in contact with a ferrous metal surface, a non-ferrous metal surface, or a combination thereof.

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Non-limiting examples of suitable metals are:

iron, copper, aluminium, magnesium, zinc, cobalt, tin and mixtures and/or alloys thereof.

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According to a fourth aspect of the present invention there is provided a kit comprising

(a) an acyl amino acid derivative of a formula (III)compound:

$$R^4 - (CO) - N(R^5) - CH_2 - CO_2H$$
, and

(III)

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(b) a N,N'-disubstituted aminomethyl triazole derivative of a formula (IV) compound:

$$T^2$$
-CH₂-NR⁶₂.

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The kit may comprise components (a) and (b) in such a ratio, that when mixed together, the resulting combination is soluble in water, organic hydrocarbons or a mixture thereof.

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The kit may further comprise a solvent comprising water and/or organic hydrocarbon or a mixture thereof.

- According to a fifth aspect of the present invention there
 is provided a method of producing a formulation comprising
 contacting:
 - (a) an acylamino acid derivative of formula (III),

15 $R^4 - (CO) - N(R^5) - CH_2 - CO_2H$, and

(III)

(b) a N, N¹-disubstituted aminomethyl triazole derivative
 of formula (IV),

 $T^2-CH_2-NR^6_2$.

(IV)

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Suitably, (a) and (b) are contacted in such a ratio that the resulting mixture is soluble in water, organic hydrocarbons or a mixture thereof.

Suitably, (a) and (b) are contacted by mixing with stirring at an elevated temperature. Temperatures from room temperature to 100°C may be employed, but preferably the temperature is kept below 50°C.

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Mixing times may vary from a few minutes to several hours. Suitably, the mixing time is at least 1 minute, preferably at least 5 minutes and more preferably at least 10 minutes. Suitably, the mixing time is less than 3 hours, preferably less than 2 hours and more preferably less than 1 hour. Suitably, mixing times are between 10-30 minutes.

The mixing step may further include addition of a solvent. 10 Suitably, a solvent is added to the compounds before stirring at an elevated temperature. Alternatively, or in addition, a solvent may be added after the process of stirring at an elevated temperature. Alternatively, or in addition, a solvent is added during the process of 15 stirring at an elevated temperature. Suitable solvents include water, organic hydrocarbons or a mixture thereof.

According to a sixth embodiment of the present invention there is provided use of a formulation according to one of 20 the preceding aspects of the invention as a corrosion inhibitor, as a rust inhibitor, as a metal passivator, as a metal deactivator, or as a multipurpose additive for a combination of the aforementioned purposes. formulation may also be used as an emulsifier or as a surfactant.

Suitably, the formulation is used in a concentration sufficient to be effective in inhibiting corrosion. The effective levels of formulation will vary depending on the 30 application. Suitably, concentrations of between 0.000001wt% and 5wt% are used, preferably concentrations of between 0.05wt% and 2wt% are used.

The use according to the sixth embodiments of the invention may be in:

hydrocarbon, synthetic and water-based hydraulic fluids,
gear oils, chain oils, circulating oils, turbine oils,
crankcase oils, compressor oils, bearing lubricants, wire
drawing lubricants, soluble oils, oil-based metalworking
fluids, metalworking fluid emulsions, grinding fluids,
heat transfer oils, electrical insulating oils, greases,
brake fluids, fuels, engine coolants, refrigeration
lubricants, surface cleaners, fountain solutions, aircraft
de-icing fluids, de-watering fluids, penetrating fluids,
polishes, adhesives, water-based paints, industrial
cooling water systems, and solutions used in the
electronics industry for printed circuit board
manufacture.

According to a seventh embodiment of the invention there is provided a method of inhibiting corrosion of a metal comprising contacting a formulation according to the first aspect of the invention and a fluid, which fluid contacts a metal susceptible to corrosion.

Any aspect of the invention may be combined with any one or more other aspect of the invention.

Formulations in accordance with the present invention find utility in a wide range of industrial oil, fuel, water, emulsion-based and surface coating systems where corrosion protection of a range of one or more metals is required.

Examples

The following examples further illustrate, but do not limit, the present invention. Unless otherwise indicated, parts and percentages are by weight.

5 Sample formulations in accordance with the present invention were prepared. Formulations outside the present invention where also prepared for comparative purposes.

The formulations were then tested for solubility and corrosion inhibition properties.

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Examples 1-15

Details of example formulations in accordance with the invention are given in Table 1 as Examples 1-15.

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The formulations of Examples 1-15 were prepared by mixing the appropriate amounts of compounds of type (III) and compounds of type (IV) in a 100ml glass conical flask and warming at 40-50°C, with stirring, for approximately 15 minutes.

Triazole derivative A is water-soluble commercial product containing bis (hydroxyethyl) aminomethyl tolutriazole isomers, sold by Ciba Specialty Chemicals under the trade name Irgamet 42 (RTM). It includes 25wt% of water.

Triazole derivative B is Irgamet 42 (RTM) as referred to above, after substantially all of the water has been removed by vacuum stripping up to 50°C.

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Examples 16 and 17 illustrate preparation of a composition according to the second aspect of the invention.

Example 16

9.33 parts of water-soluble triazole derivative A are dissolved in 150 parts of de-ionised water and the solution warmed to 40°C, with stirring in a 250ml beaker. 14.13 parts of N-oleoyl sarcosine, sold by Ciba Speciality Chemicals, are then added to the solution of triazole derivative A over approximately 10 minutes. The mixture is stirred at 40°C for a further 10 minutes and allowed to cool. The resulting product is a clear, pale yellow solution containing 13.5% solids.

Example 17

15 10.09 parts of N-oleoyl sarcosine are dissolved in 150 parts of toluene and the solution warmed to 40°C, with stirring in a 250ml beaker. 5.0 parts triazole derivative A are then added over approximately 10 minutes. The mixture is stirred at 40°C for a further 10 minutes and allowed to cool. The resulting product is a clear, pale yellow solution containing 9.1% solids.

Comparative Examples 18-22

25 Details of the formulations of Comparative Examples 18-22 are given in Table 2:

The formulations of Examples 18-22 are prepared by mixing the appropriate amounts of carboxylic acid and base in a 100ml glass conical flask and warming at 40-50°C, with stirring, for approximately 15 minutes. In these Examples, either Compound (III) or Compound (IV) of the

invention is replaced by closely related compounds outside the invention.

Solubility Testing

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The results of solubility testing of Examples 1-15 and Comparative Examples 18-22 in water and in naphthenic oil, are shown in Table 3. The results of solubility testing of Examples 2, 3, 6 and 8 in other solvents are shown in Table 4.

To test the solubility of a formulation, solutions of each formulation having a range of concentrations for each of Examples 1-15 and Comparative Examples 18-22 of, for example, 0.5%, 1.0%, 5.0%, 20% were prepared. Preparation involved taking the relevant amount of each Example formulation, mixing it with the appropriate amount of solvent to get the desired concentration and warming the solutions with stirring at 40-50°C, for approximately 15 minutes.

The solutions were then allowed to stand overnight at room temperature and then visually inspected to see if complete solution had occurred. For example, solubilities described as > 1.0% mean that 1.0% solutions are completely soluble. Solubilities described as < 1.0% meant that 1.0% solutions are not completely soluble.

Examination of Tables 3 and 4 shows that formulations

according to invention show solubility in both water and
naphthenic oil. The results also show formulations in
accordance with the invention exhibit solubility in

solvents with polarities ranging from water to paraffinic mineral oil.

The comparative examples do not show dual solubility, being soluble either in organic hydrocarbon or water, or neither, but not both.

Corrosion Inhibitor/Metal Passivator Testing

The results of testing of formulations in accordance with the invention for corrosion inhibition are given in Table 5. Details of the test methods are given below.

Ferrous metal corresion testing

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(i) Coupon Test

A 70mm x 40mm mild steel test coupon is cleaned by polishing with silicon carbide abrasive paper and degreased by immersion in toluene for one hour. The coupon is dried in the oven at 90°C. The clean, dry coupon is place in a 100ml beaker containing 50ml of a solution of the Example formulation comprising a 0.2% solution of the Example formulation in de-ionised water. The beaker is covered with a petri dish and placed in a water bath at 90°C for 1 hour. The coupon is then

removed, allowed to dry at room temperature and inspected for signs of corrosion or staining. Examples where the coupons show visual signs of corrosion or staining are not

30 considered effective corrosion inhibitors.

Examination of Table 5 shows that test solutions containing formulations in accordance with the invention

give coupons with no change over the initial clean and polished coupons. By comparison, a blank test with deionised water gives a coupon with a severe staining and corrosion showing that formulations according to the invention have corrosion inhibitor properties for mild steel.

(11) IP 287 Test (Institute of Petroleum Test Designation IP 287/82)

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This method measures the rust prevention characteristics of aqueous solutions or emulsions by the chip/filter paper method.

Cast iron chips are washed with acetone and dried in the oven at 105°C. The chips are sieved onto a filter paper placed in a petri dish so as to cover a 35mm square with a single layer. 2ml of a test solution prepared by dissolving the example formulation in synthetic hard water 20 (0.3449 g/l calcium sulphate dihydrate in distilled water) is then pipetted onto the chips so that they are all thoroughly wetted. The dish is covered with a lid and allowed to stand at room temperature for 2 hours. filter paper is then removed, washed with water and allowed to dry. The area of staining on the paper is 25 assessed. The test is repeated at various dilutions to determine the point where there is significant increase in stained area. This dilution is known as the break point and is expressed as the dilution (ratio of water to product) at the break point. 30

Examination of Table 5 shows that the inventive formulation of Example 3 gives a break point of 30 : 1

22

proving that the mixture has rust inhibiting properties for cast iron.

Aluminium corrosion testing

A 70mm x 40mm aluminium test coupon is cleaned by immersion in toluene for one hour and then dried in the oven at 90°C. The clean, dry coupon is placed in a 100ml beaker containing 50ml of test solution comprising 0.2% of an example formulation according to the invention in deionised water. The beaker is covered with a petri dish and placed in a water bath at 90°C for 20 minutes. The coupon is then removed, allowed to dry at room temperature and inspected for signs of corrosion or staining.

15 Examples where the coupons show visual signs of corrosion or staining are not considered effective corrosion

Examination of Table 5 shows that test solutions

containing formulations in accordance with the invention give coupons with no change over the initial clean and polished coupons. By comparison, a blank test with deionised water gives a coupon with severe staining and corrosion showing that formulations in accordance with the invention show corrosion inhibitor properties for aluminium.

Copper passive testing

inhibitors.

30 Copper Strip Tarnish ASTM D 130-75

The test is designed to measure the corrosiveness to copper of the hydrocarbons. Details are given in the

that all the the the that that and also that the that are the time the that that

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Annual Book of ASTM Standards published by the American Society for Testing and Materials. A clean, dry copper strip is placed in 30ml test solution comprising 0.1% of an example formulation in accordance with the invention in mineral oil containing 100ppm polysulphide and heated for 3 hours at 100°C. Results are determined by comparison with the ASTM Copper Strip Corrosion Standards and giving the appropriate classification.

- Examination of Table 5 shows that the formulation of Example 3 has a la classification in both of the mineral oils tested compared with blank ratings of 3b for the mineral oils (containing polysulphide) alone, showing the formulation of Example 3 to be a corrosion
- 15 inhibitor/passivator for copper.

The results in Table 5 show that formulations of the invention show activity as corrosion inhibitors and metal passivators for ferrous metals, aluminium and copper, both in aqueous and non-aqueous systems.

Table 1

Examples of Mixtures of the Invention

Example	Compound	Parts	Compound	Parts	Mole
No.	Type (III)	of	Туре	of	Ratio
		(m)	(IV)	(IV)	(III)
					(IV)
1	N-olcoyl sarcosine	10.59	Triazole Derivative A	6	1:06
2	N-olecyl sarcosine	10.69	Triazole Derivative A	6.57	1 : 0.65
3	N-oleoyl sarcosine	14.13	Triazole Derivative A	9.33	1:0.7
4	N-oleoyl sarcosine	10.59	Triazole Derivative A	6.57	1:0.75
5	N-oleoyl sarcosine	7.06	Triazole Derivative A	5.33	1 : 0.8
6	N-oleoyi sarcosine	10.59	Triazole Derivative A	8,49	1:0.85
7	N-oleoyi sarcosine	7.07	Triazole Derivative A	6	1:0.9
8	N-oleoyl sarcosine	10.09	Triazole Derivative B	5	1:0.7
9	N-oleoyl surcosine	9.2	Triazole Derivative B	5.21	1:0.8
10	N-olcoyl sarcosine	12.09	Triazole Derivative B	7.28	1:0.85
11	N-oleoyl saccosine	7.85	Triazole Derivative B	5	1:0.9
12	N-lauroyl sarcosine	5.42	Triazole Derivative A	4.66	1:0.7
13	N-lauroyl sarcosine	5.42	Triezole Derivative A	6.66	1:1
14	N-lauroyl sarcosine	10.84	Triazole Derivative B	7	1:0.7
15	N-lauroyl sarcosine	5.71	Triazole Derivative B	5.26	1:1

Table 2

Comparative Examples

Example	Carboxylle Acid	Parts of	Base	Parts of	Mole
No.		Acid		Base	Rade
					Acid/
					Base
18	N-olcoyl sarcosine	7.06	Benzotriazole	1.67	1:07
19	N-oleoyl sarcosine	7.53	Tolumazole	2	1:07
20	N-olcoyl sarcosine	7.06	Triethanolamine	2.09	1:07
21	Oleic Acid	8.08	Triazole Derivative A	6.67	1:07
22	Laurie Acid	5.75	Triazole Derivative A	6.67	1:07

Table 3
Solubility Tests in Water and Naphthenic Oil

Formulation of	Soli	ability (%)
Example No.	Water*	Naphthenic Oil*
1	ca 0.5	ca 0.3
2	ca 0.5	ca 0.3
3	> 20	ca 2.5
• 4	> 20	ca 1.0
5	> 20	ca 0.5
6	> 20	ea 0.5
7	> 20	ca 0.5
8	> 20	> 10
9	> 20	> 1.0
10	> 20	> 1.0
11	> 20	ся 1.0
12	>15	> 1.0
13	> 15	> 1.0
14	> 15	> 5.0
15	> 15	> 1.0
Comparative Example No.		
18	< 1.0	> 2.5
19	< 1.0	> 2.5
20	> 2.0	< 0.5
21	< 0.5	< 0.5
22	< 0.5	< 0.5

* Solvents:

Water

Laboratory Grade De-ionised Water

Naphthenic Oil

Nynas T 22 supplied by Nynas Naphthenics Limited

<u>Table 4</u>

<u>Solubility Tests in Additional Solvents</u>

		Solubility (%)		
Paraffinic Oll*	Gasoline*	Toluene*	PAG*	Ethylene Glycol*
			> 1.0	
	> 5.0	i	> 1.0	
1			> 1.0	
>10	> 5.0	> 10	ca 3.0	> 5.0
	Paraffinic Ou*	> 5.0	Paraffinic Oil* Gasoline* Toluene*	Paraffinic Oil* Gasoline* Toluene* PAG*

* Solvents

Paraffinic Oil # 100 Solvent Neutral Oil supplied by Mobil UK
Gasoline - Commercial 95 Octane Esso Unleaded Petrol
Toluene - 99% Laboratory Reagent Grade
PAG - Breox 50 A20 (RTM) supplied by Inspec UK Ltd

Ethylene Glycol = Laboratory Reagent Grade Ethanediol

Table 5

Corrosion Inhibitor/Metal Passivator Tests

		<u></u>			- 1
Passivation Test Rate	082*	æ	<u>s</u>		
Cu Passivation ASTM D 130 Test Rate	•1 BO	9 8	1.8		
Al Corroston	Coupon Test	Severe corrosion/staining No corrosion/staining	No corrosion/staining	No corrosion/staining	No corrosion/staining
	IP 287 Test Breakpoint		30:1		
Fe Corroston	Coupen Test	Severe corrosion/staining No corrosion/staining	No corresion/staining	No corresion/staining	No corresion/staining
Example No.		Blank (no additive)	27%	vs.	66

* Oil 1 = 100 Solvent Neutral Oil supplied by Mobil

= 500 Solvent Neutral Oil supplied by Batoyle Freedom Group

* Oil 2

The reader's attention is directed to all papers and documents which are filed concurrently with or previous to this specification in connection with this application and which are open to public inspection with this specification, and the contents of all such papers and documents are incorporated herein by reference.

All of the features disclosed in this specification (including any accompanying claims, abstract and drawings), and/or all of the steps of any method or process so disclosed, may be combined in any combination, except combinations where at least some of such features and/or steps are mutually exclusive.

15 Each feature disclosed in this specification (including any accompanying claims, abstract and drawings), may be replaced by alternative features serving the same, equivalent or similar purpose, unless expressly stated otherwise. Thus, unless expressly stated otherwise, each feature disclosed is one example only of a generic series of equivalent or similar features.

The invention is not restricted to the details of the foregoing embodiment(s). The invention extend to any novel one, or any novel combination, of the features disclosed in this specification (including any accompanying claims, abstract and drawings), or to any novel one, or any novel combination, of the steps of any method or process so disclosed.

Claims

- A formulation comprising:
- 5 a) an acyl amino acid derivative of the formula (III):

$$R^4 - (CO) - N(R^5) - CH_2 - CO_2H$$

(III)

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in which

R4 is Ca - C30 optionally substituted alkyl

and R5 is hydrogen or methyl, and

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b) a N,N'-disubstituted aminomethyl triazole derivatives of the formula (IV):

 T^2 -CH₂-NR⁶₂

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(IV)

in which

 T^2 is an optionally substituted 1,2,3-benzotriazole group,

or an optionally substituted 1,2,4-triazole group, and

R⁶ is a hydroxyalkyl group

A formulation according to Claim 1, wherein the
 formulation is soluble in organic hydrocarbons.

- 3. A formulation according to Claim 2, wherein the solubility at 25°C and 1 atmosphere pressure is at least 0.000001wt% in Naphthenic oil.
- 5 4. A formulation according to Claim 2, wherein the solubility at 25°C and 1 atmosphere pressure is at least 0.000001wt% in toluene.
- 5. A formulation according to Claim 1, wherein theformulation is soluble in water.
 - 6. A formulation according to Claim 5, wherein the solubility at 25°C and 1 atmosphere pressure is at least 0.000001wt% in water.

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- 7. A formulation according to Claim 1, wherein the formulation is soluble in both water and organic hydrocarbons.
- 20 8. A formulation according to Claim 1, wherein the mole ratio of the formula (III) compound to the formula (IV) compound is from 1: 0.2 to 1: 2.
- 9. A formulation according to Claim 1, which comprises25 further additives.
 - 10. A composition comprising:
- a) a formulation according to any one of the precedingclaims

and

b) a diluent.

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- 11. A composition according to Claim 9, wherein the diluent is water, organic hydrocarbon, or a mixture thereof.
- 5 12. A composition according to Claim 11, wherein the organic hydrocarbon comprises natural or synthetic aliphatic or aromatic compounds of carbon and hydrogen, optionally containing unsaturated linkages, ester groups or hetero atoms.

13. A composition according to Claim 11, wherein the organic hydrocarbon is selected from the group comprising:

- octane, kerosine, white spirit, petroleum-based hydrocarbons such as naphthenic oils or paraffinic oils, vegetable oils, synthetic carboxylic acid ester, phosphate esters, poly α olefins, poly isobutylenes, alkylated aromatic hydrocarbons, ethylene glycol, propylene glycol, polyalkylene glycols, glycol ethers.
 - 14. A composition according to Claim 11, wherein water is selected from the group comprising:
- 25 distilled water, de-ionised water, natural water and synthetic hard water.
- 15. A composition according to Claim 10, in contact with a ferrous metal surface, a non-ferrous metal surface or a combination thereof.
 - 16. A kit comprising

(a) an acyl amino acid derivative of formula (III):

 $R^4 - (CO) - N(R^5) - CH_2 - CO_2H$,

5 (III)

and,

(b) a N-N'-disubstituted aminomethyl triazole

derivative of formula (IV):

 $T^2-CH_2-NR^6_2$

(IV)

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- 17. A kit according to Claim 16, further comprising a solvent comprising water and/or organic hydrocarbon or a mixture thereof.
- 20 18. A kit according to Claim 16 comprising quantities of (a) and (b) in such a ratio that, when mixed together, the resulting formulation is soluble in water, organic hydrocarbon or a mixture thereof.
- 25 19. A method of producing a formulation comprising contacting:
 - (a) an acyl amino acid derivative of formula (III):

30 $R^4 - (CO) - N(R^5) - CH_2 - CO_2H$

(III)

and,

a N, N' - disubstituted aminomethyltriazol derivative of formula (IV):

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 $T^2-CH^2-NR^6_2$.

(IV)

- 20. A method according to Claim 19, wherein (a) and (b) are contacted in such a ratio that the resulting mixture is soluble in water.
- 21. A method according to Claim 19, wherein (a) and (b) 15 are contacted in such a ratio that the resulting mixture is soluble in organic hydrocarbon.
 - 22. A method according to Claim 19, wherein (a) and (b) are contacted by mixing with stirring at an elevated temperature.
 - 23. A method according to Claim 19, further comprising addition of a diluent.
- 25 24. A method according to Claim 23, wherein the diluent is added before stirring at an elevated temperature.
 - 25. A method according to Claim 23, wherein the diluent is added during stirring at an elevated temperature.

26. A method according to Claim 23, wherein the diluent is added after stirring at an elevated temperature.

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- 27. Use of a formulation according to any one of the preceding claims, as a corrosion inhibitor, as a rust inhibitor, as a metal passivator, as a metal deactivator, as an emulsifier, as a surfactant or as a multi purpose additive for a combination of the aforementioned purposes.
- 28. Use of a formulation according to Claim 27, wherein a concentration of between 0.000001wt% and 5wt% of the formulation are used.
- 29. A method in inhibiting corrosion of a metal comprising contacting a formulation according to any one of Claims 1-9 and a fluid, which fluid contacts a metal susceptible to corrosion.

IMPROVEMENTS IN AND RELATING TO CORROSION INHIBITING FORMULATIONS

Abstract

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The present invention relates to formulations for corrosion-inhibiting, emulsifying, lubricating or as a surfactant in organic hydrocarbon and/or water based system containing an acyl amino acid derivative and a triazole derivative.

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